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First Synthesis of Conagenin Diastereoisomers

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Abstract: The first synthesis of two conagenin analogs is described. Copyright © 1996 Elsevier Science Ltd

Conagenin (1), a low molecular weight immunomodulator, was discovered in fermentation broths of Streptomyces roseosporus MI696-AF3¹. It is an immunomodulator of antitumor activity stimulating activated T cells and enhancing generation of antitumor cells. As conagenin itself does not show cytotoxicity to murine and human tumor cells and was found to be effective *in vivo* in improving antitumor activity of cyclophosphamide, mitomycin C and adriamycin against murine leukemias, it may be useful in cancer chemotherapy²⁻⁵.

To our best knowledge there have been no synthetic efforts towards conagenin and its analogs yet. Herein, we wish to report the synthesis of the 3'-epimer of naturally occurring conagenin (3'-epiconagenin, 2) and its 2R-diastereoisomer (3, Fig. 1).

5-Deoxy-1,2-O-isopropylidene-α-D-xylofuranose (4), which can be prepared from D-xylose in four steps⁶, was chosen as starting material for synthesis of the dideoxy aldonic acid part of 2 and 3 (Fig. 2). Oxidation of the secondary OH-group with chromium trioxide-pyridine complex⁷ resulted in the 3-ulose 5, which was readily converted into the exomethylene compound 6⁸ via a Wittig reaction. Hydrogenolysis of

Fig. 2.

Conditions: i) CrO₃·2pyr, Ac₂O, CH₂Cl₂; ii) Ph₃PCH₃Br, nBuLi, THF, 60% from 4; iii) 10% Pd/C, H₂, EtOAc, 87%; iv) EtSH, cc.HCl, THF, 0°C, 84%; v) BnBr, NaH, Bu₄NI, THF, 97%; vi) HgCl₂, CdCO₃, acetone/water, vii) PDC, DMF, 55% from 9; viii) HOBT, DCC, (S)-O-benzyl-2-methylserine methyl ester¹¹, CH₂Cl₂, 83%; ix) 1M KOH, dioxane, 82%; x) 10% Pd/C, H₂, MeOH, 66%

6 with 10% Pd on charcoal catalyst gave the 3-deoxy-3-C-methyl derivative 7 as a single diastereomer, which was identical (by ¹H-NMR and specific rotation) with the compound synthesized by Tronchet et al. via a different route. Direct mercaptalization of 7 gave the D-ribo-derivative 8, which was efficiently benzylated via conventional methodology. Mercury salt promoted demercaptalization of 9 afforded unstable aldehyde 10, which was oxidized without purification. For this purpose the Corey method¹⁰ proved to be the most effective.

The syrupy ribonic acid derivative 11 was coupled with (S)-O-benzyl-2-methylserine methyl ester¹¹ by the use of HOBT and DCC giving rise to the fully protected conagenin diastereomer 12. The ester group was hydrolized with 1M KOH solution in dioxane. Subsequent catalytic hydrogenation of compound 13 led to 3'-epiconagenin (2).

Using (R)-O-benzyl-2-methylserine methyl ester¹² the same procedures were followed to obtain 3 (Fig. 3).

Conditions: xi) HOBT, DCC, (R)-O-benzyl-2-methylserine methyl ester¹², CH₂Cl₂, 77%; xii) 1M KOH, dioxane, 82%; xiii) 10% Pd/C, H₂, MeOH, 95%

Fig. 3

The synthesis of conagenin and its other analogs is in progress in our laboratory. The biological activity of 2, 3 and further analogs compared with 1 will be published later.

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- 8. All of the compounds gave satisfactory microanalytical and/or spectroscopic evidence. Selected spectroscopic and physical data are the following. Compounds without mp are syrupy substances. Specific rotations were measured at room temperature. Compound 6: $[\alpha]_0 = +127.1$ (c 0.63 CHCl₃); ¹H NMR (200 MHz, CDCl₃) δ 5.08 (1H, d) and 5.33 (1H, d, C=CH₂, J= 2 Hz). Compound 7: $[\alpha]_D$ = +33.3 (c 0.98, CHCl₃); lit.⁸ +36.1 (c 1.2, CHCl₃). Compound 8: $[\alpha]_p = -49.2$ (c 2.25 CHCl₃); FAB-MS m/z 238 (M⁺). Compound 9: $[\alpha]_D = +15.1$ (c 1.07, CHCl₃); EI-MS m/z 357 (M+H-SEt)⁺. Compound 10: FTIR (KBr) $v_{c=0} = 1732 \text{ cm}^{-1}$; EI-MS m/z 283 (M-CHO)⁺. Compound 11: $[\alpha]_{p} = +3.4$ (c 0.71, CHCl₂); FTIR (KBr) $v_{c=0} = 1714 \text{ cm}^{-1}$; EI-MS m/z 329 (M+H)*. Compound 12: $[\alpha]_p = +20.0$ (c 0.69, CHCl₃); FAB-MS m/z 534 (M⁺). Compound 13: mp 152-153 °C (CH₂Cl₂-hexane); $[\alpha]_p = +4.0$ (c 0.62, CHCl₃); CI-MS m/z 520 (M⁺). Compound 2: (2S)N-[(2'R,3'R,4'R)2',4'-dihydroxy-3'-methylpentanoyl]-2methylserine, mp 105-106 °C (MeOH, dec.); [α]_p= +25.8 (c 1.25, MeOH); MS (thermospray) m/z 250 $(M+H)^+$; ¹H NMR (400 MHz, MeOD): δ 0.94 (3H, d, $J_{3:3\text{-methyl}} = 7.1$ Hz, 3'-CH₃), 1.16 (3H, d, $J_{4:5} =$ 6.4 Hz, 5'-H), 1.43 (3H, s, 2-CH₃), 1.95 (1H, m, 3'-H), 3.82 (1H, dq, J_{V,4}= 7.1 Hz, 4'-H), 3.83 and 4.04 (2H, ABq, J=10.9 Hz, CH_2OH), 4.08 (1H, d, $J_{2/3}=4.0$ Hz, 2'-H); ¹³C NMR (90 Hz, MeOD): δ 180.5 and 175.9 (carbonyl and amide), 75.0 (2'-C), 69.8 (4'-C), 66.7 (CH₂OH), 63.1 (2-C), 45.8 (3'-C), 21.4 (5'-C), 20.5 (2-CH₃), 13.1 (3'-CH₃). Compound 14: $[\alpha]_D = +22.7$ (c 0.93, CHCl₃); EI-MS m/z 535 $(M+H)^{+}$. Compound 15: mp 137-138 °C ($CH_{2}Cl_{2}$ -hexane); $[\alpha]_{D}$ = +22.0 (c 0.75, $CHCl_{3}$); EI-MS m/z 428 (M-Bn)*. Compound 3: (2R)N-[(2'R,3'R,4'R)2',4'-dihydroxy-3'-methylpentanoyl]-2-methylserine, mp 168-170 °C (MeOH, subl.); $[\alpha]_D$ = +6.0 (c 1.06, MeOH); FAB-MS m/z 250 (M+H)⁺; ¹H NMR (200 MHz, D_2O): δ 0.92 (3H, d, $J_{3',3'-\text{methyl}} = 7.1$ Hz, 3'-CH₃), 1.17 (3H, d, $J_{4',5'} = 6.4$ Hz, 5'-H), 1.42 (3H, s, 2-CH₃), 2.01 (1H, m, 3'-H), 3.83 and 3.94 (2H, ABq, J= 11.3 Hz, CH_2OH), 3.91 (1H, dq, $J_{3'4}$ = 7.1 Hz, 4'-H), 4.10 (1H, d, $J_{2:3}$ = 4.8 Hz, 2'-H); ¹³C NMR (90 MHz, D_2 O): 180.2 and 175.4 (carbonyl and amide), 74.9 (2'-C), 69.2 (4'-C), 65.9 (CH₂OH), 63.2 (2-C), 44.4 (3'-C), 20.3 (5'-C), 20.2 (2-CH₃), 12.3 $(3'-CH_3).$
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